

The Degradation Mechanisms of Matisse and van Gogh's Pigments – Probing Photo-oxidation Reactions at the Nanoscale



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**Matisse's *Le Bonheur de vivre* (*The Joy of Life*, 1905-6)
the single most consequential early modern painting
many of Picasso's works competitive response**

four versions - yellows fading, darkening,
spalling in Barnes version

mottled surface -
tan, ochre hues

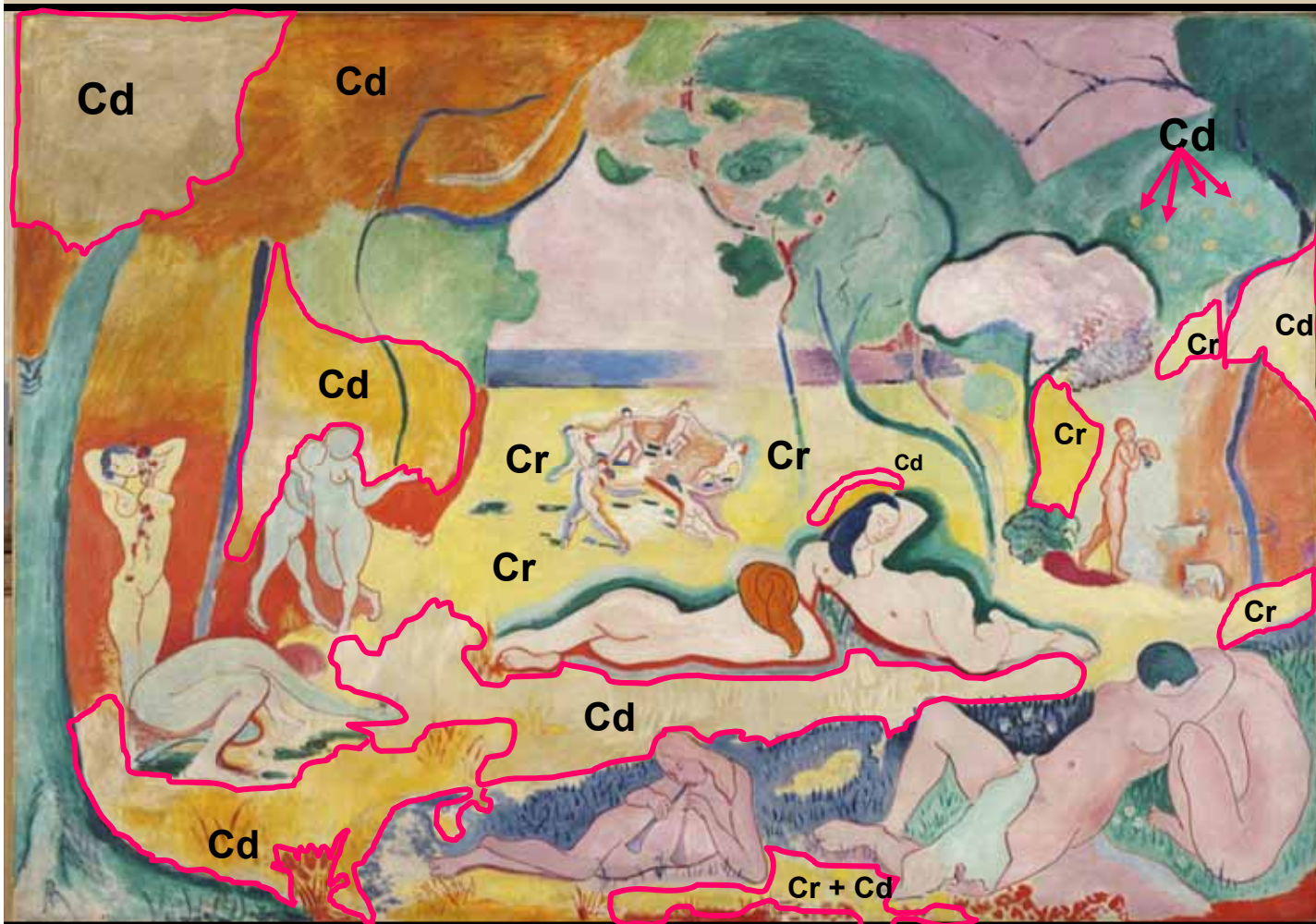


S. F. MOMA final oil
sketch (1905-6)



Severe
degradation of
yellow paint -
fading, flaking,
spalling

Handheld XRF Survey Overview



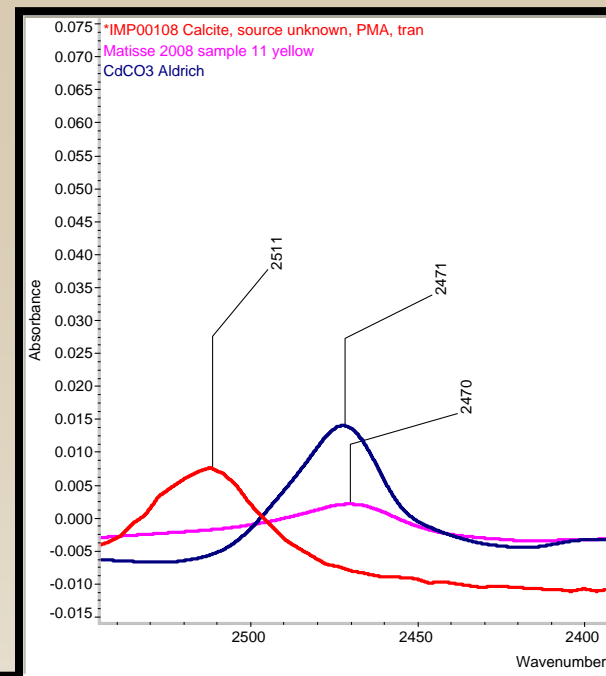
Degradation confined to Cd yellows

CdS-based pigments

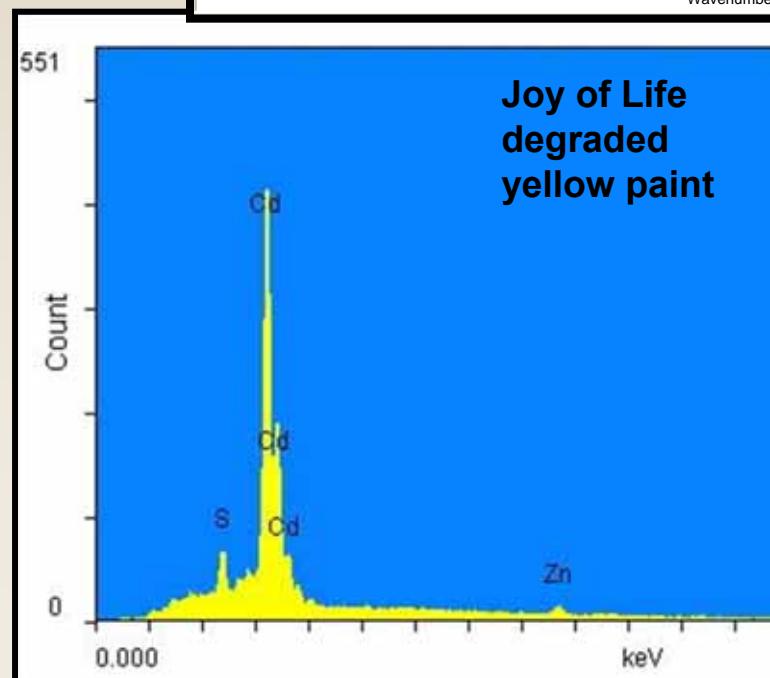
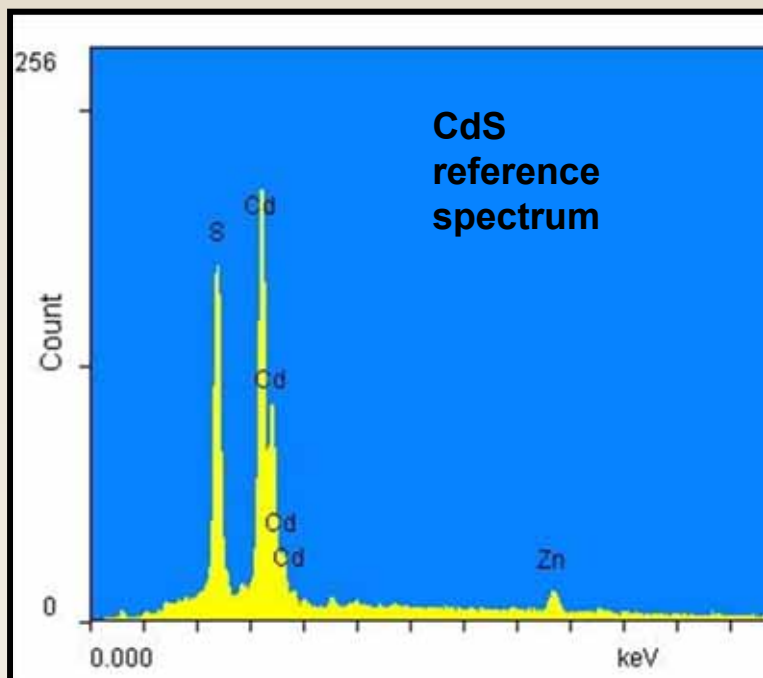


Elemental and Molecular Analysis Degraded Yellows – Confirm CdS?

- SEM-EDS
 - Stoichiometry incorrect
 - CdS no longer major Cd phase
- X-Ray Diffraction
 - CdCO_3 only
 - No CdS powder pattern
 - α -CdS (greenockite), β -CdS (hawleyite)
 - CdS, if present, is amorphous or nanocrystalline
- FTIR
 - CdCO_3



(B. Berrie, G. Gates, M. Palmer NGA)



Synthesis of CdS Using Period Recipes for Matisse Paint Replication

- **XRD, HR-TEM used to monitor**

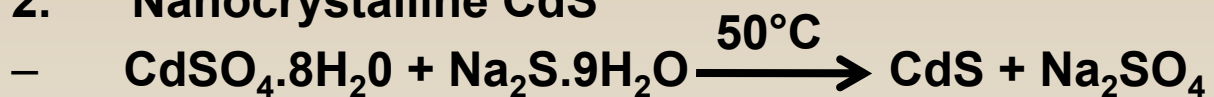
1. **Amorphous CdS**



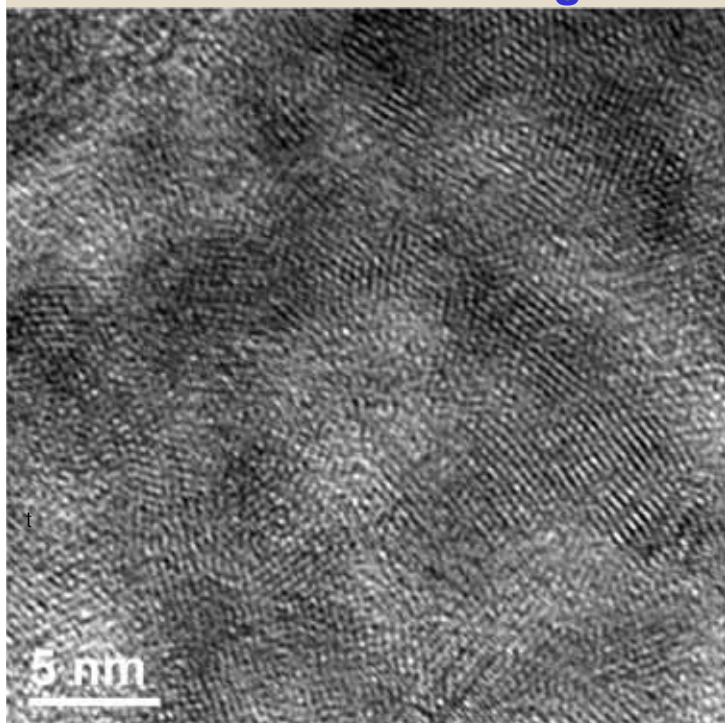
- Translucent ivory solid

- **Would not have been used as a yellow pigment**

2. **Nanocrystalline CdS**

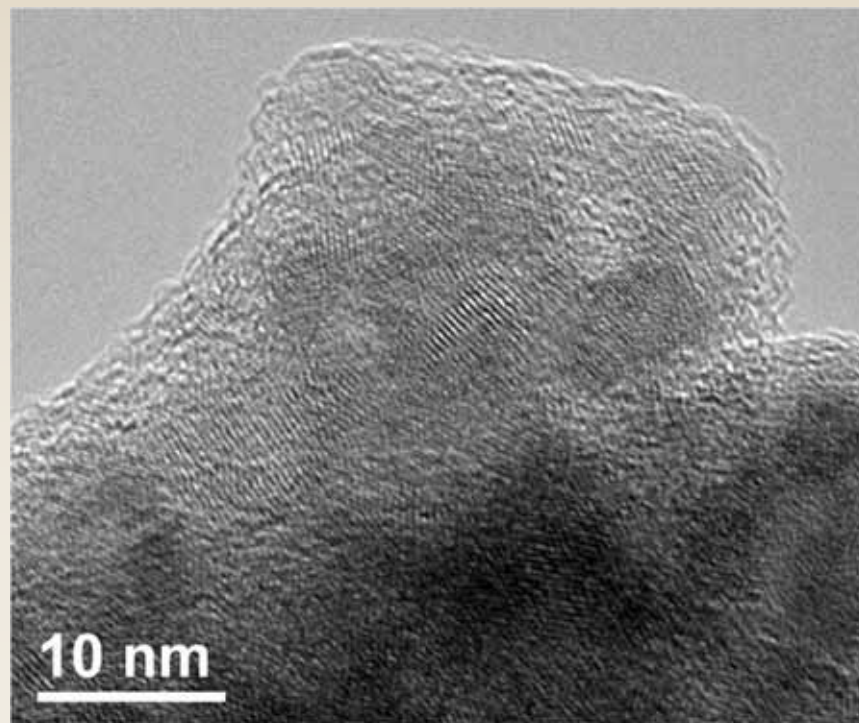


- **Yellow-orange solid - 2.9 nm crystallite size**



No
calcination
step in
turn of 20th
c.

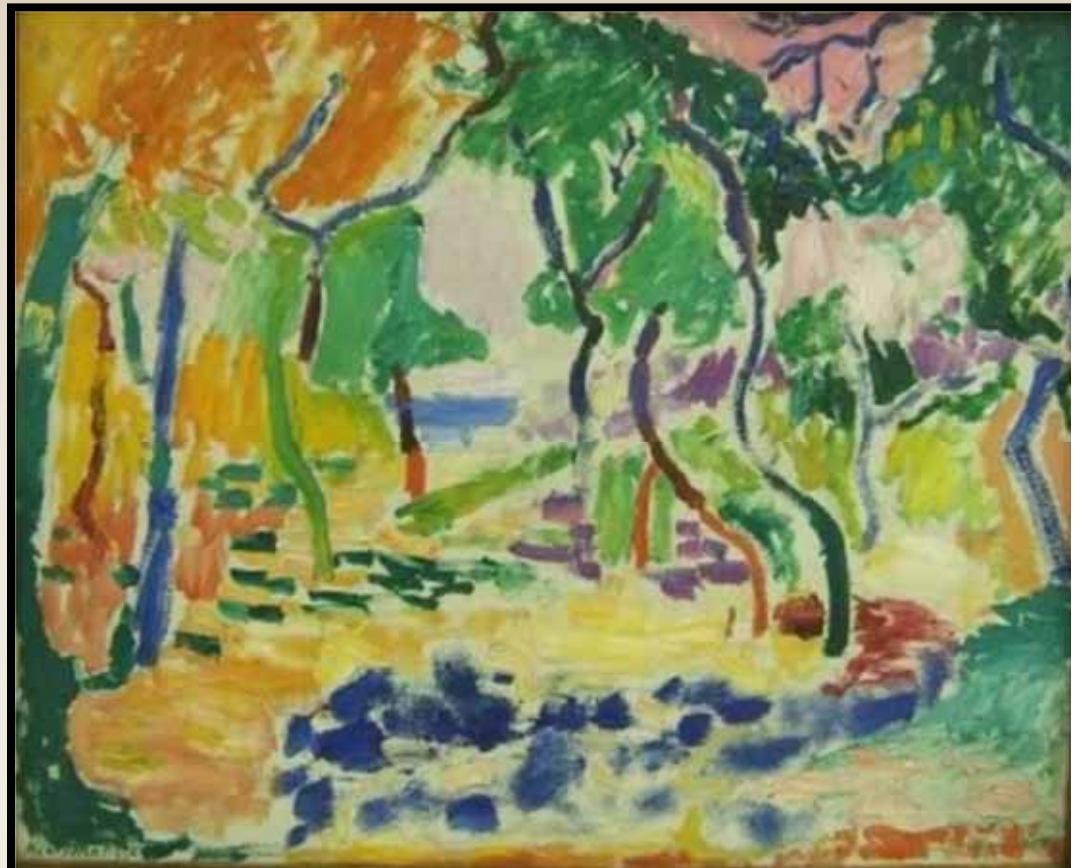
caused
later
reactivity



One Explanation for CdCO_3 : Photo-Oxidative Degradation of CdS

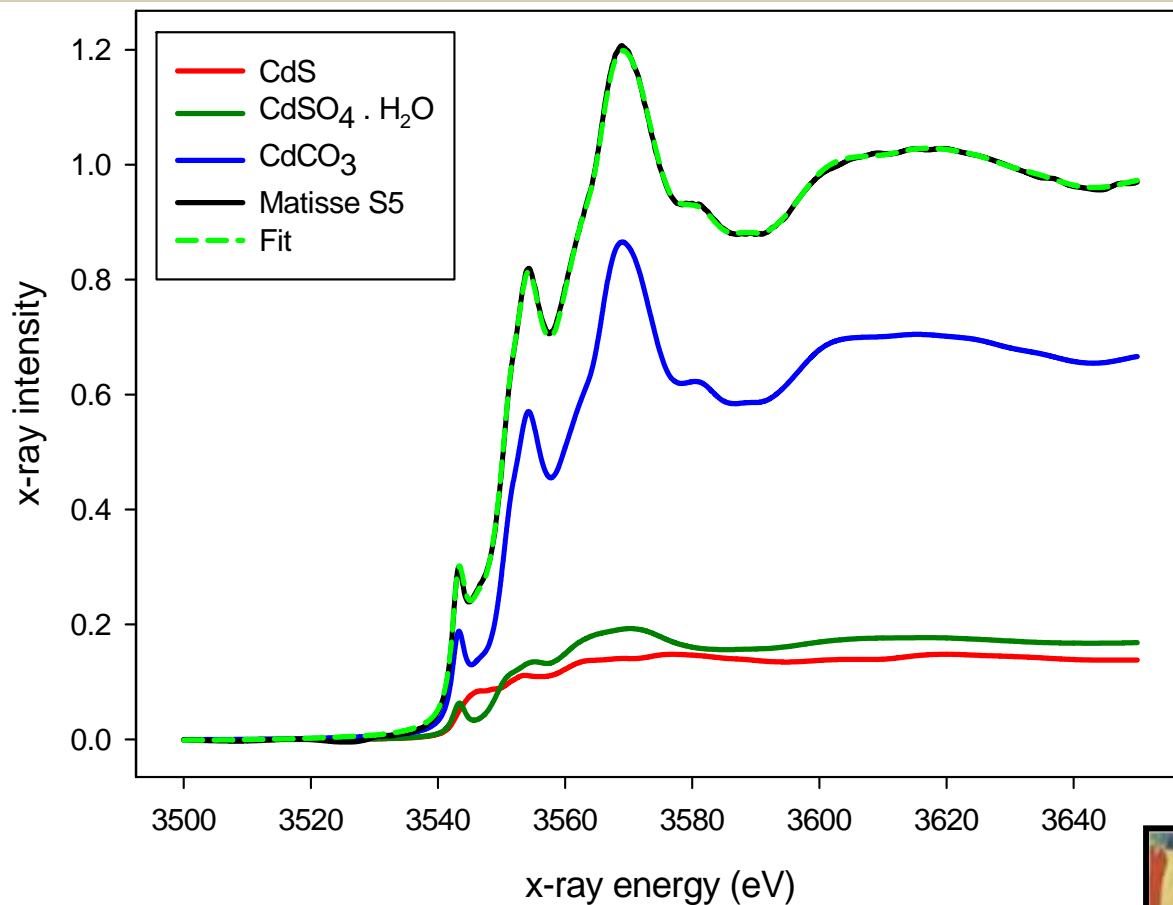
- When photon E incident light \approx band gap (2.42 eV, 512 nm)
 - $\text{CdS} + h\nu = \text{CdS} + e^- + h^+$
 - $\text{CdS} + 2h^+ = \text{Cd}^{2+} + \text{S}(s)$
 - $\text{CdS} + 2\text{O}_2 = \text{Cd}^{2+} + \text{SO}_4^{2-}$
 - $\text{Cd}^{2+} + \text{SO}_4^{2-} = \text{CdSO}_4$
 - $\text{CdS} + 1.5 \text{O}_2 = \text{CdO} + \text{SO}_2$
 - $\text{CdO} + \text{CO}_2 = \text{CdCO}_3$

Landscape at Collioure/Study for
Le Bonheur de vivre (1905)
Statens Museum for Kunst
Copenhagen



Cd L Edge XANES Data

10 micron
beam size



Best fit local composition:

68.6 % CdCO₃

14.2 % CdS

17.1 % CdSO₄ · nH₂O

Consistent with XRD,
FTIR

CdSO₄ · nH₂O makes
proposed photo-
oxidation mechanism
more likely

opportunity to re-evaluate scholarship of iconic work

“...the dirty yellow ground is for the most part patchily
scrubbed in...”

Matisse and the Subject of Modernism, Alastair Wright,
2004



Georges Seurat *A Sunday on La Grande Jatte* (1884) The Art Institute of Chicago (Casadio et al.)



First Campaign: May 1884 - March 1885

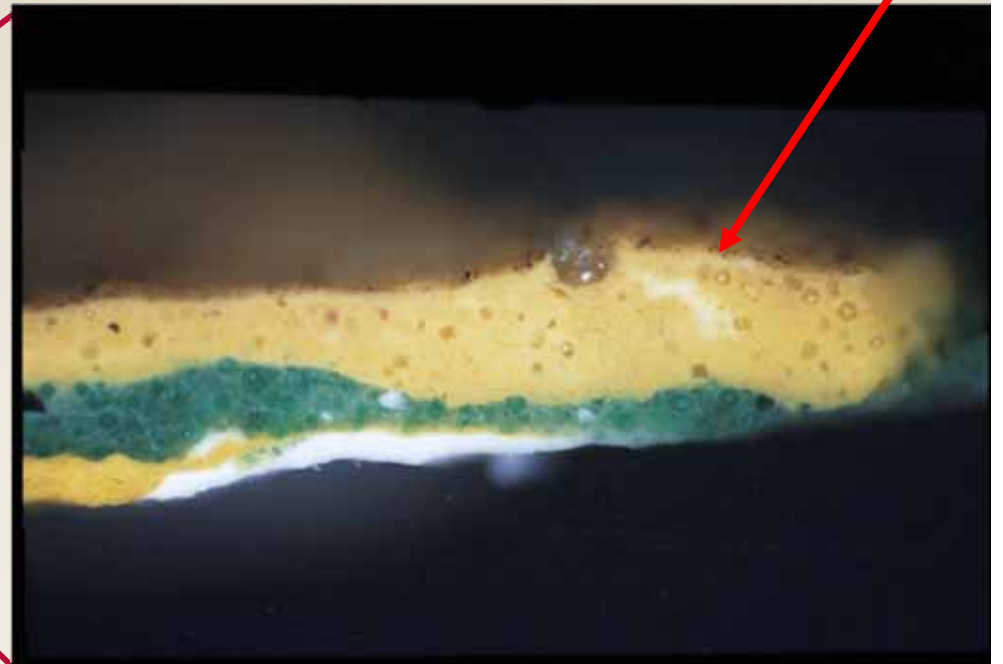
Second Campaign: October 1885 - March 1886

Third Campaign: 1888/1889

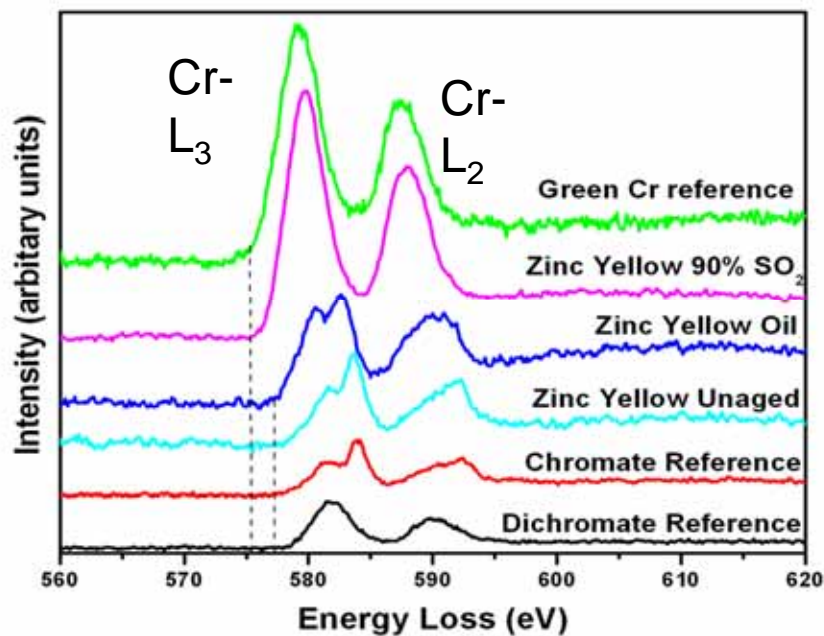
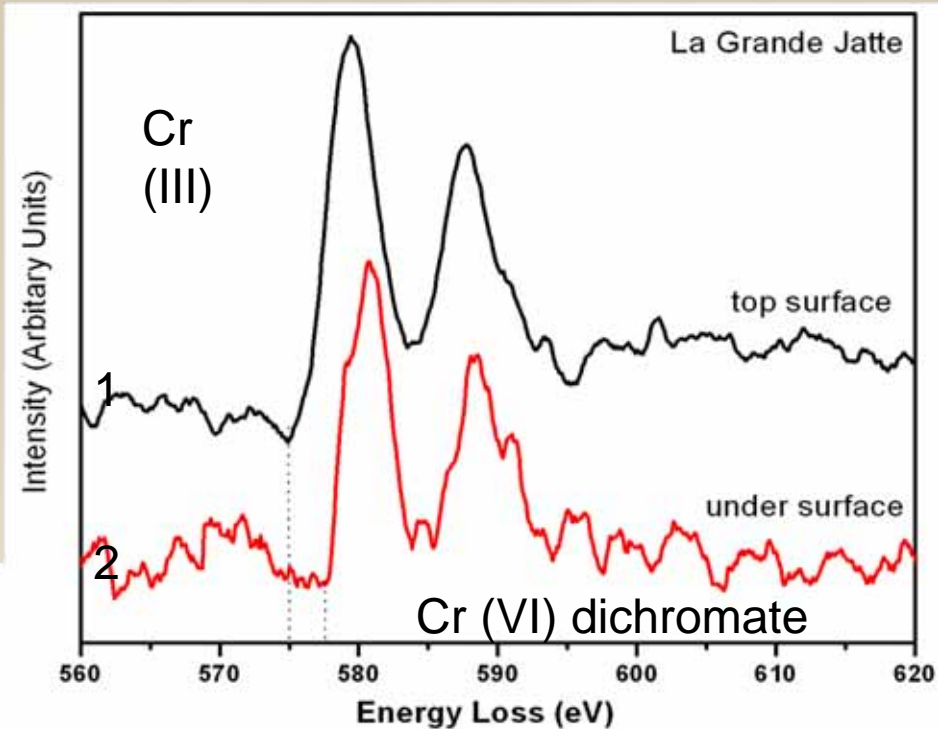
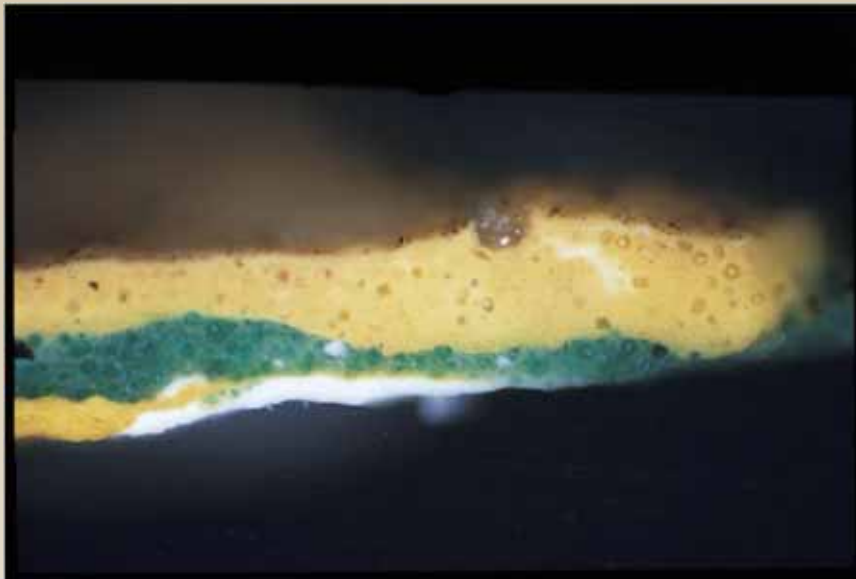
L. Zanella et al., "The Darkening of zinc yellow: XANES speciation of chromium in artist's paints after light and chemical exposures", *J. Anal.At. Spectrom.* 26 (2011) 1090-1097.



- **zinc yellow** ($\text{K}_2\text{O} \cdot 4\text{ZnCrO}_4 \cdot 3\text{H}_2\text{O}$)
- bright greenish-yellow
- upon manufacture
- darkens rapidly to dull, ochre yellow hue
- photochemical reduction of chromates?
- greenish color of zinc yellow due to transformation of chromate (CrO_4^{2-}) into chromium oxide (Cr_2O_3)?



Cr EELS on Altered Zinc Yellow from La Grande Jatte



Cr (III)

Cr (VI)

PRESENT STATE

AFTER DIGITAL REJUVENATION

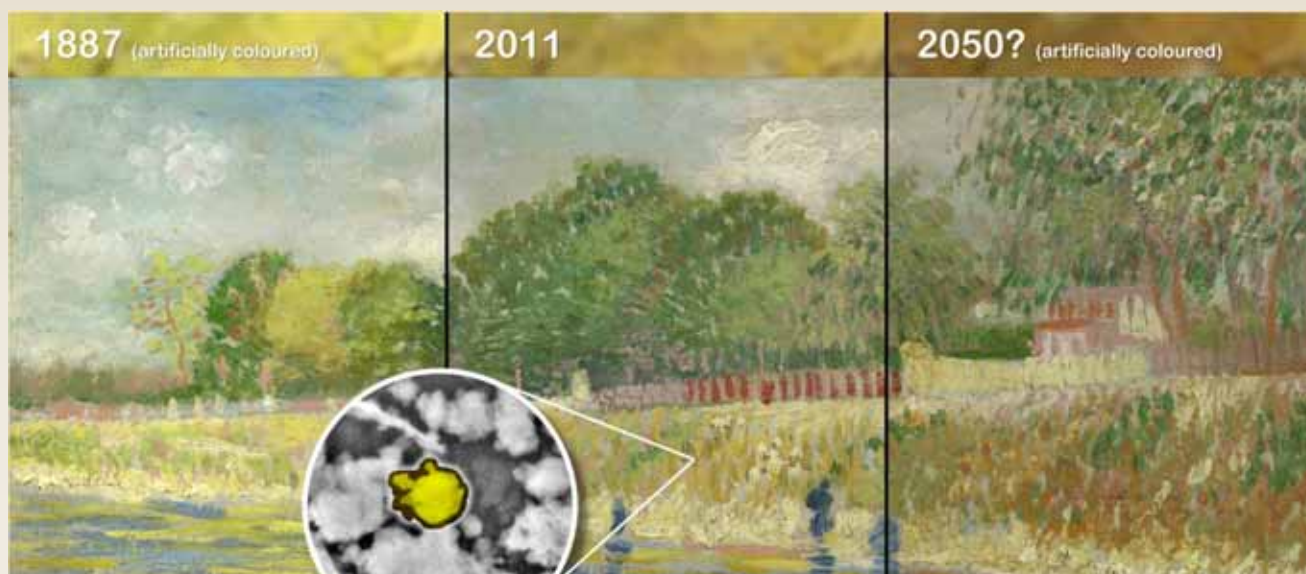


**PbCrO₄
photo-
reduction
in works by
van Gogh**

*Bank of the
Seine (V.
Van Gogh,
1887)*



J. Dik,
Koen
Janssens,
et al.



Images from:
[http://www.vangogh.
ua.ac.be/](http://www.vangogh.ua.ac.be/)

The Bedroom (October 1888)
when the artist was living in the Yellow House in Arles

Photomicrographs of
altered PbCrO_4 (chrome
yellow) paint

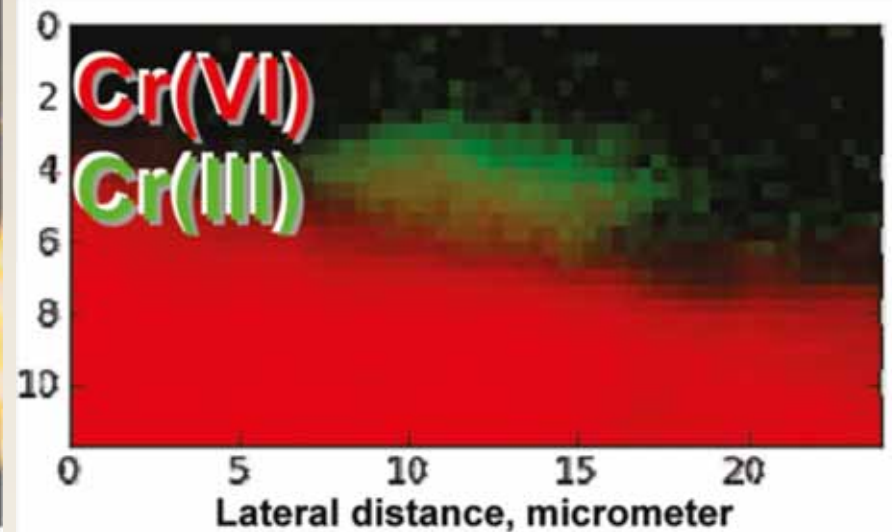
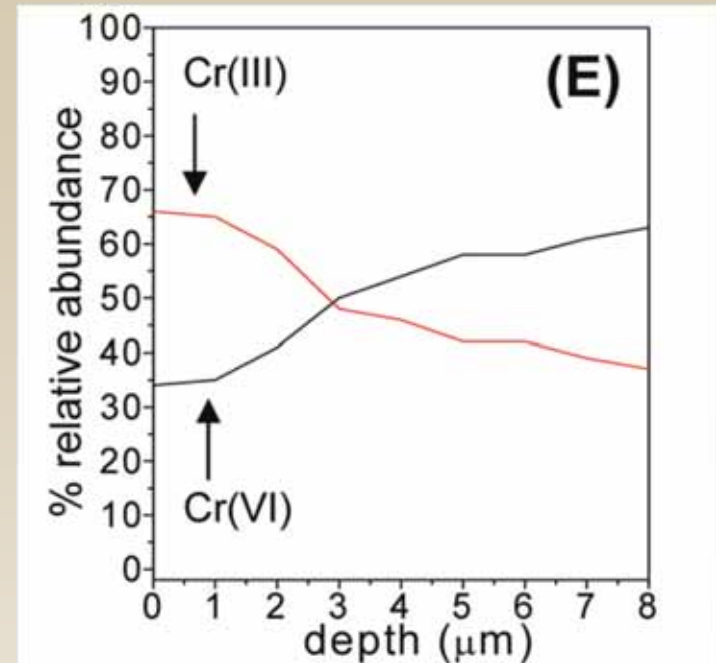
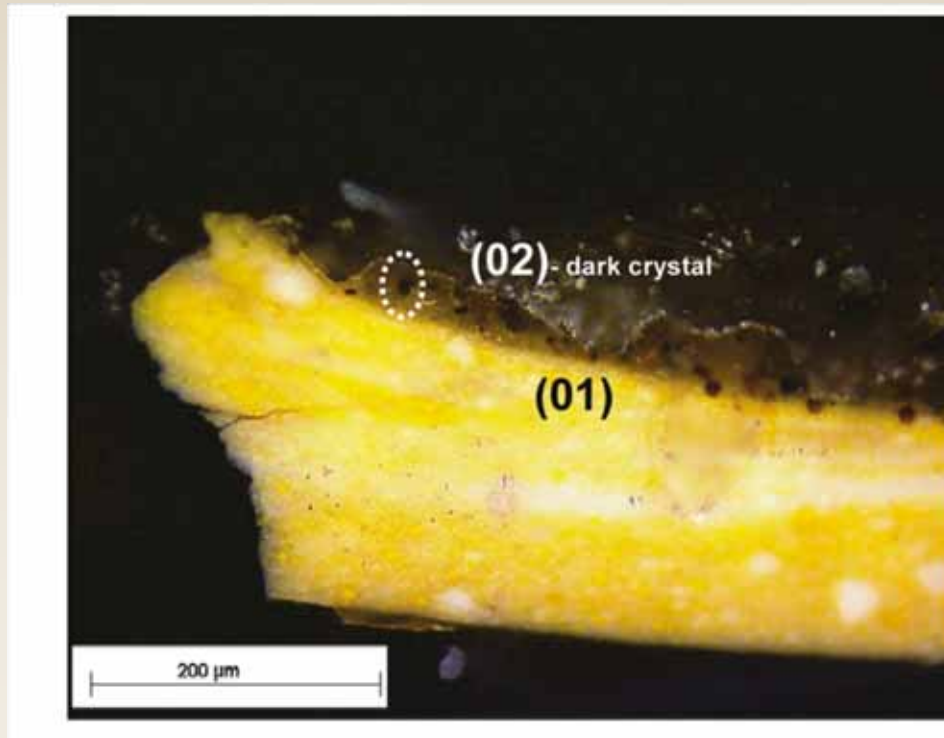
Alteration layer less
than $3 \mu\text{m}$ thick



<http://www.vangogh.ua.ac.be/>

μ XANES Mapping

"Degradation Process of Lead Chromate in Paintings by Vincent van Gogh Studied by Means of Synchrotron X-ray Spectromicroscopy and Related Methods. 2. Original Paint Layer Samples", by Letizia Monico et al., *Analytical Chemistry* 83 (2011) 1224-1231,



Proposed ERL or USR Experiments – Diffraction Limited Hard X-Ray Source

- **Objective: identification of incipient photodegradation in Impressionist and Early Modern paintings**
- **Alteration occurring in the top nanometers of paint layers**
 - **Damage visible/disfiguring when 1- 3 μm of photodegradation reached**
- **Allows preventive conservation methods before damage is visible**
 - Argon-filled cases
 - Controlled light levels
 - RH control
- **Speciation of paint surfaces as a function of depth:**
 - **CdS photo-oxidation reactions**
 - $\text{CdSO}_4 \cdot n\text{H}_2\text{O}$
 - CdO
 - CdCl_2 – leftover starting reagent decreased band gap
 - $\text{Cd}(\text{OH})_2$
 - CdCO_3
 - **PbCrO_4 and $\text{K}_2\text{O} \cdot 4\text{ZnCrO}_4 \cdot 3\text{H}_2\text{O}$ photoreduction**
 - Cr_2O_3
 - PbCrO_4
 - $\text{K}_2\text{O} \cdot 4\text{ZnCrO}_4 \cdot 3\text{H}_2\text{O}$



Experiments for Determining Speciation as a Function of Depth or Photoaging for Model Systems

Applications of nanoscale x-ray probes

- nanoXANES – speciation mapping in paint cross-sections
 - “screen” painting collections for evidence of incipient photodegradation
 - MOMA, Getty, Barnes, MMA, Whitney, Art Institute of Chicago, National Gallery
- Confocal nanoXANES – on intact paintings
 - Directly probe within and beneath painting’s photodegraded “skin”
 - Unmounted paint flakes
 - Avoid solubility issues
- Time-resolved XANES on artificially aging model samples
- Time-resolved XRD on artificially aging samples

Spatially resolved 3D micro-XANES by a confocal detection scheme

Geert Silversmit, Bart Vekemans, Sergey Nikitenko, Sylvia Schmitz, Tom Schoonjans, Frank E. Brenker and Laszlo Vincze
Phys. Chem. Chem. Phys., 2010, **12**, 5653-5659

18.5 x 12.0 x 10.0 μm^3 at the Cu K-edge

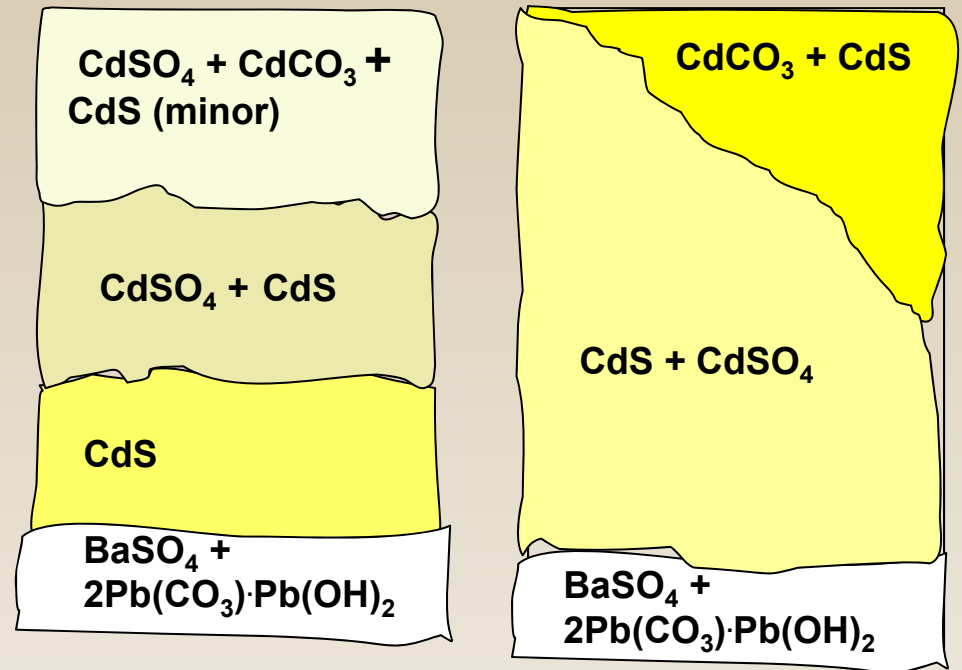
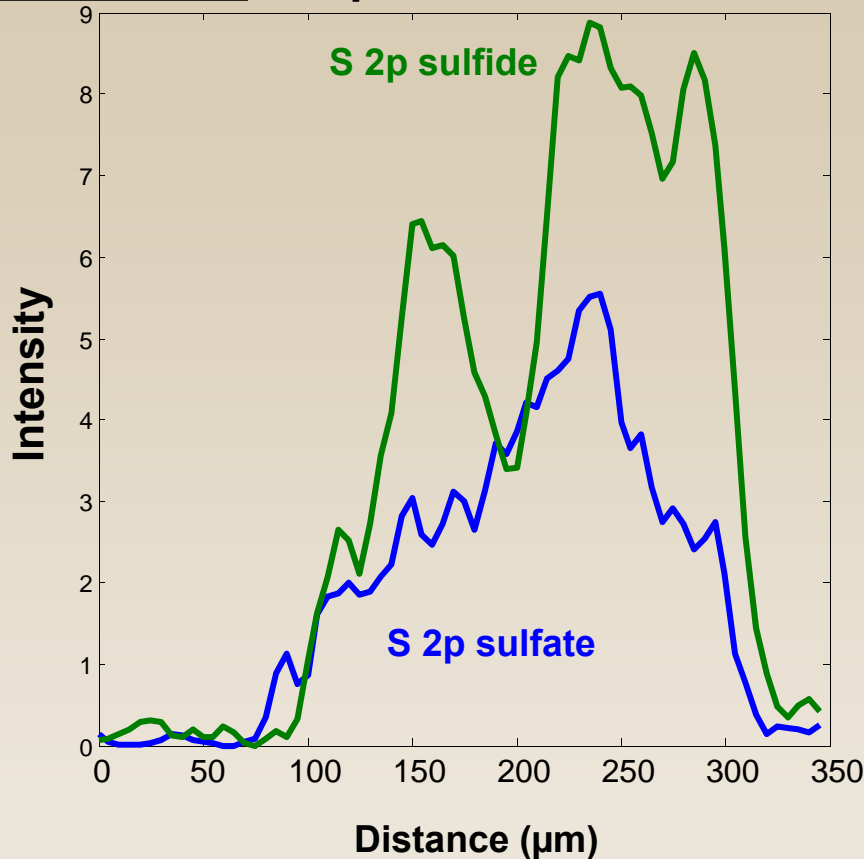




Why Pigment Alteration Problems Require Resolution Beyond Microbeams (limitations of cross-sections):

XPS S²⁻ and SO₄²⁻ Concentrations
Cross Section Line Scan - 10 micron steps

S 2p line scans



SO₄²⁻ peaks in center of x-section

CdCO₃ dominates surface chemistry

phase distribution does not fit photo-oxidation model
proposed – solubility and transport phenomena reflected
instead?

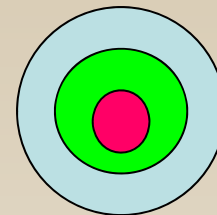


“Coring” of Individual Pigment Particles

Objective: Photodegradation mechanisms obtained

- impact of transport phenomena eliminated
- impact of solubility phenomena eliminated

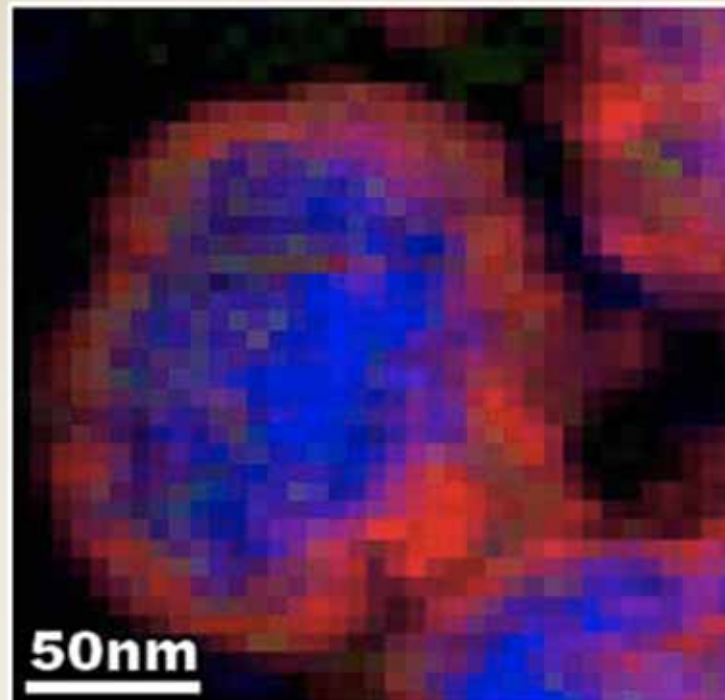
CdS pigment particle
size – 1-20 μm



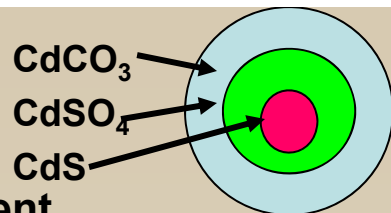
STEM/EELS map of chromium oxide (Cr_2O_3) from one of the grains of PbCrO_4 inside the brown alteration layer

newly formed Cr(III) compound - a
nanometers-thin coating of the pigment
particles

"Degradation Process of Lead Chromate in Paintings by Vincent van Gogh Studied by Means of Synchrotron X-ray Spectromicroscopy and Related Methods. 1. Artificially Aged Model Samples",
by Letizia Monico, Geert Van der Snickt, Koen Janssens, Wout De Nolf, Costanza Miliani, Johan Verbeeck, He Tian, Haiyan Tan, Joris Dik, Marie Radepont, and Marine Cotte,
Analytical Chemistry
83 (2011) 1214-1224



nanoXRF Experiments

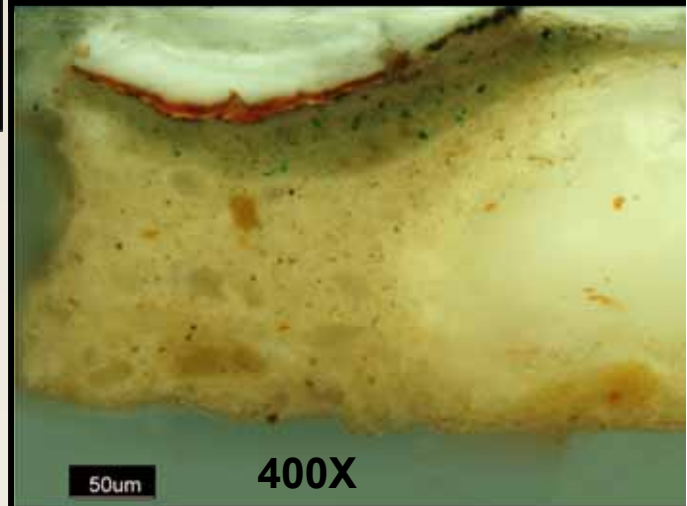
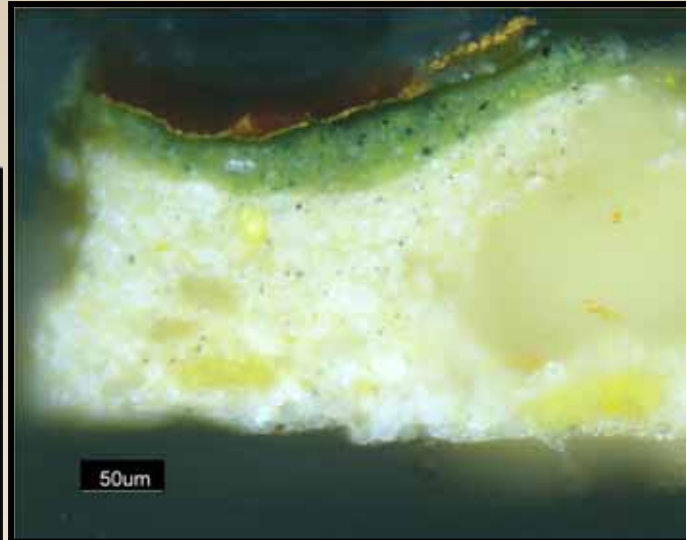
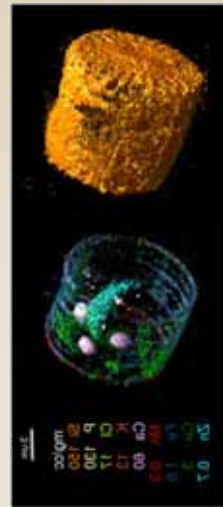


- Cd/S as a function of depth in pigment particles
 - traditional geometry for paint cross-sections
 - photoaged model systems
 - microsamples removed from paintings
 - confocal or conventional geometry for “XRF tomography”



Trends in hard X-ray fluorescence mapping: environmental applications in the age of fast detectors, E. Lombi et al., **Analytical and Bioanalytical Chemistry**, 400(6) 1637-1644, JUN 2011

- confocal geometry for intact paintings



nanoTomography Experiments

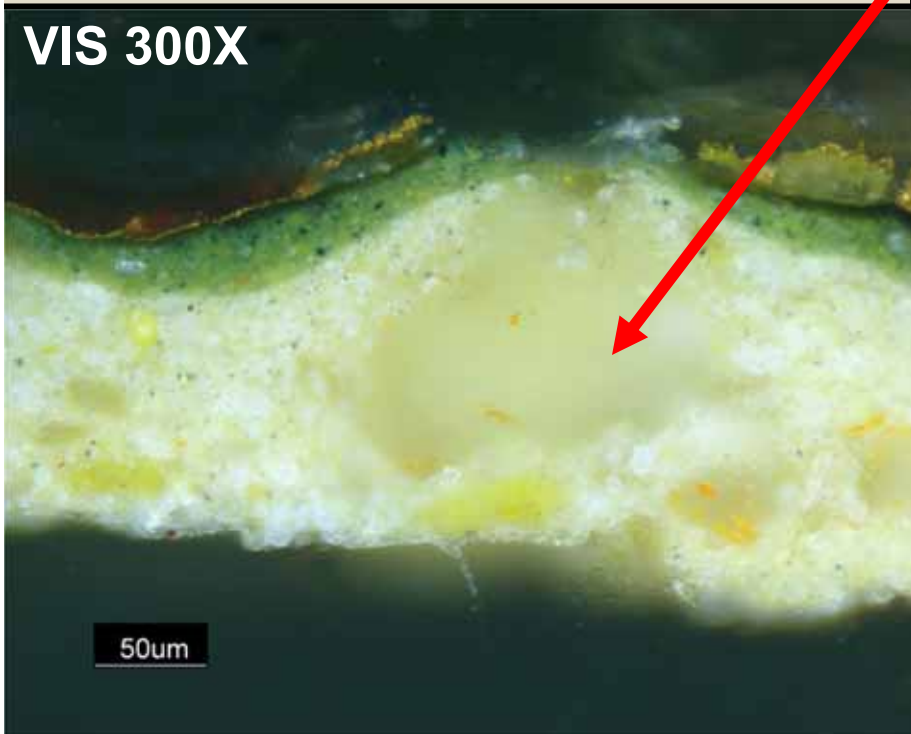
- nanoscale x-ray tomography of paint cross-sections
- imaging of alteration zones, reaction zones
- imaging of materials migrating through cross-sections (J. Boon μm length scale)
- growth of lead carboxylate aggregates
- growth of new phases inside aggregates (Pb_3O_4)

Lead Soap Aggregates

- lead pigments react with free fatty acids in drying oil
 - Form lead carboxylate salts/soaps
- phase separate from rest of layer
- Creates pebbly surface
- pinpoint paint losses
- monitor formation by nanoscale x-ray scattering?



VIS 300X



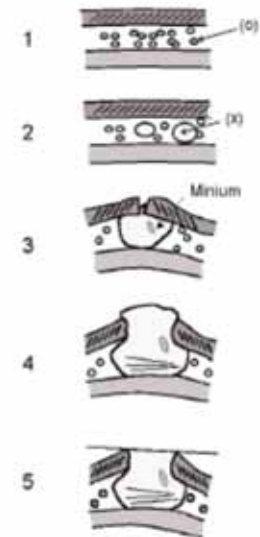
+ H₂O



Free fatty acids come from hydrolysis of paint layers

Slow formation and aggregation by controlling R.H.

Protrusion dynamics:



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